

# ANALOX

## HYP O2 Portable User Manual

**Analox Ltd.**

15 Ellerbeck Court, Stokesley Business Park  
North Yorkshire, TS9 5PT, UK

**T:** +44 (0)1642 711400 **F:** +44 (0)1642 713900  
**W:** [www.analox.net](http://www.analox.net) **E:** [info@analox.net](mailto:info@analox.net)

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## 1 Introduction

### SAFETY WARNING

Please read the information provided in Section 7 regarding the chemicals used inside the oxygen sensor. Normally, these are retained inside the sensor and pose no risk. However, if due to a fault they leak out of the sensor, the analyser must then be handled with care as detailed.

The HYP O2 Portable is a versatile portable analyser, suitable for monitoring partial pressure oxygen in several applications, including:

- Diving bells
- Chamber atmospheres
- Hyperbaric lifeboats
- Submarines

The compact size allows for convenient fitting within confined spaces such as diving bells and hyperbaric lifeboats, or alternatively the unit can easily be hand carried.

It is very simple to use, with just a single push-button to switch power on and off, and an LCD display. A single 9V PP3 battery will power the analyser continuously for approximately 800 hours (1 month). A single recessed potentiometer permits calibration when necessary.

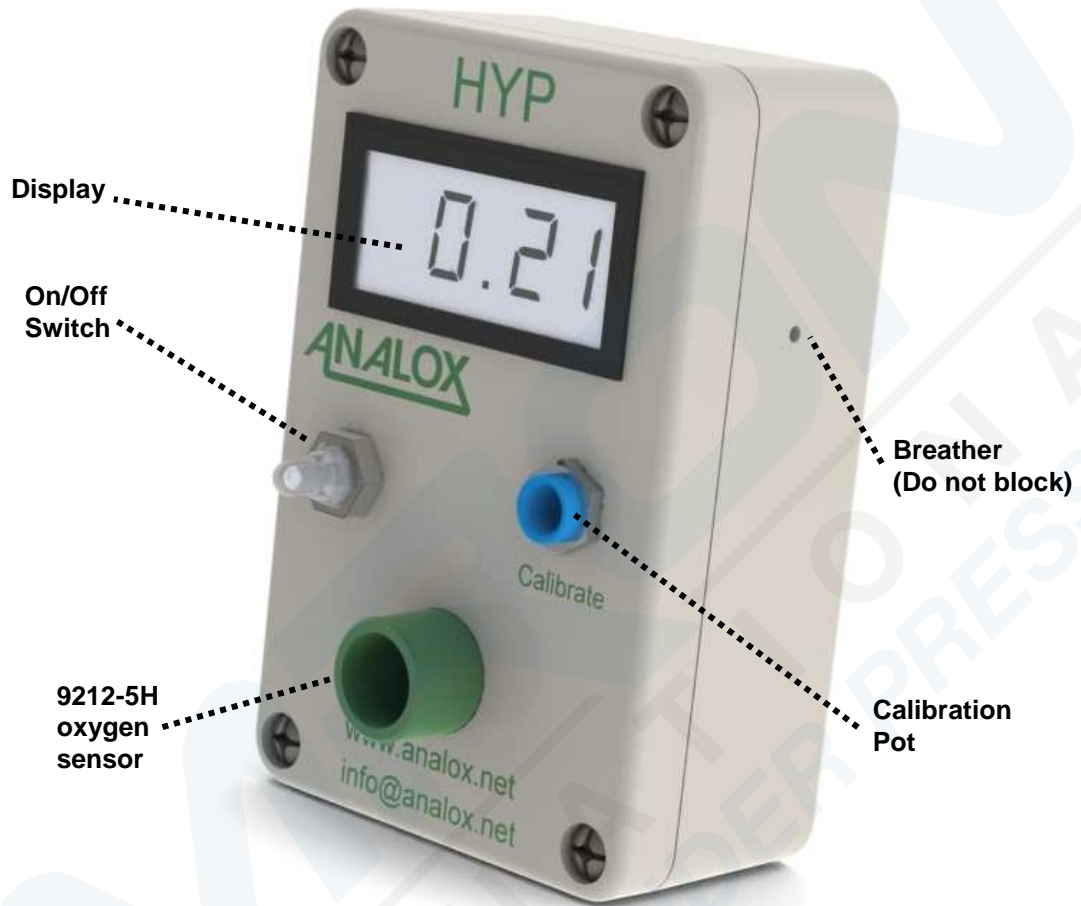
It may be used at depths of up to 600MSW / 2000FSW, and can also be used at surface pressures.

All circuit boards are specially coated to ensure protection against humidity and condensation. Readings are provided in units of bar.

The HYP O2 Portable uses the latest electrochemical cell technology to ensure a fast response, stable reading and long life (up to 3 years in 20.9% oxygen at RTP). All sensors are individually temperature compensated to optimise performance across the full operating temperature range.

The analyser may be used on its own for monitoring the local atmosphere directly, or alternatively it may be used with a sampling kit available as an accessory. This allows samples to be drawn from remote areas for measurement.

2 Overview



### 3 Operation

#### 3.1 Switching on and off

Simply switch the analyser on or off by switching the on/off switch. The oxygen measurement will appear on the LCD screen within one or two seconds.

The oxygen reading will be in the units specific to the analyser. This manual assumes that bar are the units of display.

#### 3.2 Ambient/Local atmosphere monitoring

When monitoring the local atmosphere around the analyser, gas will diffuse into the oxygen sensor from the atmosphere, and the display will indicate the partial pressure of oxygen to which it has been exposed. Avoid breathing out over the sensor, since exhaled breath contains a significant proportion of carbon dioxide and a corresponding deficiency of oxygen which will affect the accuracy of the reading. Conversely, some customers use the effect of breathing on the sensor as a confidence check prior to taking a reading, just to make sure that the analyser is functioning. However, they must then wait (usually say around 30 seconds) until normal mixing of the atmosphere allows the reading to settle back to a true reading.

#### 3.3 Remote monitoring

To monitor a remote atmosphere, the analyser may be used in conjunction with the remote sampling kit. Note it is assumed that the remote area is at the same pressure as the location in which the analyser is being used. The sampling kit consists of an aspirator bulb which allows the user to hand-pump a sample of gas from the remote area through a sample line and a flow adaptor which is pushed into the oxygen sensor inlet port. The aspirator must be pumped enough times to flush the sample through the sample line. As supplied, the sample line is 2m long, but this can be extended.

#### 3.4 Low battery warning

When the analyser is in use and the battery is almost exhausted (terminal voltage less than approximately 5.5V) a battery symbol will appear in the top left corner of the display. In this condition, the analyser will continue to measure oxygen for a few hours, but it is advisable to replace the battery as soon as possible. As the battery depletes further, the oxygen reading will ultimately be affected. Replace the battery as detailed in Section 5.1.



### 3.5 Note on partial pressure measurements

The analyser measures the partial pressure of oxygen. This is not to be confused with the percentage of oxygen in the atmosphere.

In an environment where it is the partial pressure of oxygen that is physically being maintained (for example a diving bell), then the readout from the analyser is simple to understand.

So for example, in a diving bell being maintained at 400mbar ppO<sub>2</sub> the analyser will simply read 0.4bar ppO<sub>2</sub>. This will be the same whatever the depth of the diving bell, provided that the ppO<sub>2</sub> is maintained correctly.

In a submarine being maintained at a nominal air concentration of 20.9% O<sub>2</sub>, the reading on the analyser will vary according to the actual pressure inside the submarine. At 1bar absolute pressure, 20.9% oxygen will equate to a partial pressure of oxygen of 0.21bar

Using the formula

$$\text{Oxygen ppO}_2 \text{ (bar)} = \text{Oxygen Percentage} * \text{Absolute Pressure (bar)} / 100$$

allows a table of ppO<sub>2</sub> values at various pressures to be created, assuming that the atmosphere is maintained at 20.9% O<sub>2</sub>. Table 1 can be used when you know the atmospheric pressure in bar.

**Table 1: ppO<sub>2</sub> (bar) of 20.9% O<sub>2</sub> at x bar Atmospheric Pressure**

Example: at a pressure of 1.05 Bar, 20.9% oxygen content gives a partial pressure of 0.22 bar ppO<sub>2</sub>

barA	0	0.01	0.02	0.03	0.04	0.05	0.06	0.07	0.08	0.09
0.8	0.17	0.17	0.17	0.17	0.18	0.18	0.18	0.18	0.18	0.19
0.9	0.19	0.19	0.19	0.19	0.20	0.20	0.20	0.20	0.20	0.21
1	0.21	0.21	0.21	0.22	0.22	0.22	0.22	0.22	0.23	0.23
1.1	0.23	0.23	0.23	0.24	0.24	0.24	0.24	0.24	0.25	0.25
1.2	0.25	0.25	0.25	0.26	0.26	0.26	0.26	0.27	0.27	0.27
1.3	0.27	0.27	0.28	0.28	0.28	0.28	0.28	0.29	0.29	0.29
1.4	0.29	0.29	0.30	0.30	0.30	0.30	0.31	0.31	0.31	0.31
1.5	0.31	0.32	0.32	0.32	0.32	0.32	0.33	0.33	0.33	0.33
1.6	0.33	0.34	0.34	0.34	0.34	0.34	0.35	0.35	0.35	0.35
1.7	0.36	0.36	0.36	0.36	0.36	0.37	0.37	0.37	0.37	0.37
1.8	0.38	0.38	0.38	0.38	0.38	0.39	0.39	0.39	0.39	0.40
1.9	0.40	0.40	0.40	0.40	0.41	0.41	0.41	0.41	0.41	0.42
2	0.42	0.42	0.42	0.42	0.43	0.43	0.43	0.43	0.43	0.44
2.1	0.44	0.44	0.44	0.45	0.45	0.45	0.45	0.45	0.46	0.46
2.2	0.46	0.46	0.46	0.47	0.47	0.47	0.47	0.47	0.48	0.48
2.3	0.48	0.48	0.48	0.49	0.49	0.49	0.49	0.50	0.50	0.50
2.4	0.50	0.50	0.51	0.51	0.51	0.51	0.51	0.52	0.52	0.52
2.5	0.52	0.52	0.53	0.53	0.53	0.53	0.54	0.54	0.54	0.54
2.6	0.54	0.55	0.55	0.55	0.55	0.55	0.56	0.56	0.56	0.56
2.7	0.56	0.57	0.57	0.57	0.57	0.57	0.58	0.58	0.58	0.58
2.8	0.59	0.59	0.59	0.59	0.59	0.60	0.60	0.60	0.60	0.60
2.9	0.61	0.61	0.61	0.61	0.61	0.62	0.62	0.62	0.62	0.62
3	0.63	0.63	0.63	0.63	0.64	0.64	0.64	0.64	0.64	0.65
3.1	0.65	0.65	0.65	0.65	0.66	0.66	0.66	0.66	0.66	0.67
3.2	0.67	0.67	0.67	0.68	0.68	0.68	0.68	0.68	0.69	0.69
3.3	0.69	0.69	0.69	0.70	0.70	0.70	0.70	0.70	0.71	0.71
3.4	0.71	0.71	0.71	0.72	0.72	0.72	0.72	0.73	0.73	0.73
3.5	0.73	0.73	0.74	0.74	0.74	0.74	0.74	0.75	0.75	0.75
3.6	0.75	0.75	0.76	0.76	0.76	0.76	0.76	0.77	0.77	0.77
3.7	0.77	0.78	0.78	0.78	0.78	0.78	0.79	0.79	0.79	0.79
3.8	0.79	0.80	0.80	0.80	0.80	0.80	0.81	0.81	0.81	0.81
3.9	0.82	0.82	0.82	0.82	0.82	0.83	0.83	0.83	0.83	0.83
4	0.84	0.84	0.84	0.84	0.84	0.85	0.85	0.85	0.85	0.85
4.1	0.86	0.86	0.86	0.86	0.87	0.87	0.87	0.87	0.87	0.88
4.2	0.88	0.88	0.88	0.88	0.89	0.89	0.89	0.89	0.89	0.90
4.3	0.90	0.90	0.90	0.90	0.91	0.91	0.91	0.91	0.92	0.92
4.4	0.92	0.92	0.92	0.93	0.93	0.93	0.93	0.93	0.94	0.94
4.5	0.94	0.94	0.94	0.95	0.95	0.95	0.95	0.96	0.96	0.96
4.6	0.96	0.96	0.97	0.97	0.97	0.97	0.97	0.98	0.98	0.98
4.7	0.98	0.98	0.99	0.99	0.99	0.99	0.99	1.00	1.00	1.00
4.8	1.00	1.01	1.01	1.01	1.01	1.01	1.02	1.02	1.02	1.02
4.9	1.02	1.03	1.03	1.03	1.03	1.03	1.04	1.04	1.04	1.04
5	1.05	1.05	1.05	1.05	1.05	1.06	1.06	1.06	1.06	1.06
5.1	1.07	1.07	1.07	1.07	1.07	1.08	1.08	1.08	1.08	1.08
5.2	1.09	1.09	1.09	1.09	1.10	1.10	1.10	1.10	1.10	1.11
5.3	1.11	1.11	1.11	1.11	1.12	1.12	1.12	1.12	1.12	1.13
5.4	1.13	1.13	1.13	1.13	1.14	1.14	1.14	1.14	1.15	1.15
5.5	1.15	1.15	1.15	1.16	1.16	1.16	1.16	1.16	1.17	1.17
5.6	1.17	1.17	1.17	1.18	1.18	1.18	1.18	1.19	1.19	1.19
5.7	1.19	1.19	1.20	1.20	1.20	1.20	1.20	1.21	1.21	1.21
5.8	1.21	1.21	1.22	1.22	1.22	1.22	1.22	1.23	1.23	1.23
5.9	1.23	1.24	1.24	1.24	1.24	1.24	1.25	1.25	1.25	1.25
6	1.25	1.26	1.26	1.26	1.26	1.26	1.27	1.27	1.27	1.27
6.1	1.27	1.28	1.28	1.28	1.28	1.29	1.29	1.29	1.29	1.29
6.2	1.30	1.30	1.30	1.30	1.30	1.31	1.31	1.31	1.31	1.31
6.3	1.32	1.32	1.32	1.32	1.33	1.33	1.33	1.33	1.33	1.34
6.4	1.34	1.34	1.34	1.34	1.35	1.35	1.35	1.35	1.35	1.36
6.5	1.36	1.36	1.36	1.36	1.37	1.37	1.37	1.37	1.38	1.38
6.6	1.38	1.38	1.38	1.39	1.39	1.39	1.39	1.39	1.40	1.40
6.7	1.40	1.40	1.40	1.41	1.41	1.41	1.41	1.41	1.42	1.42
6.8	1.42	1.42	1.43	1.43	1.43	1.43	1.43	1.44	1.44	1.44
6.9	1.44	1.44	1.45	1.45	1.45	1.45	1.45	1.46	1.46	1.46
7	1.46	1.47	1.47	1.47	1.47	1.47	1.48	1.48	1.48	1.48

### 3.6 Conversion of partial pressure measurement to a percentage measurement

To calculate the percentage concentration of oxygen from the reading of partial pressure, the user will need to know the actual pressure in the location at which the measurement has been taken.

So if the reading on the analyser is R bar ppO<sub>2</sub>  
and the actual pressure is known to be P bar (absolute),  
then the %O<sub>2</sub> = 100 x R / P

Table 2 may be used to do this calculation. Note the blank part of the table which shows that the ppO<sub>2</sub> cannot exceed the actual pressure.

**Table 2: Calculate %O<sub>2</sub> from ppO<sub>2</sub> and Pressure in bar**

ppO <sub>2</sub> (bar)	0.10	0.20	0.30	0.40	0.50	0.60	0.70	0.80	0.90	1.00	1.10	1.20	1.30	1.40	1.50	1.60	1.70	1.80	1.90	2.00	
Pressure (bara)																					
0.80	12.5	25.0	37.5	50.0	62.5	75.0	87.5	100.0													
1.00	10.0	20.0	30.0	40.0	50.0	60.0	70.0	80.0	90.0	100.0											
1.20	8.3	16.7	25.0	33.3	41.7	50.0	58.3	66.7	75.0	83.3	91.7	100.0									
1.40	7.1	14.3	21.4	28.6	35.7	42.9	50.0	57.1	64.3	71.4	78.6	85.7	92.9	100.0							
1.60	6.3	12.5	18.8	25.0	31.3	37.5	43.8	50.0	56.3	62.5	68.8	75.0	81.3	87.5	93.8	100.0					
1.80	5.6	11.1	16.7	22.2	27.8	33.3	38.9	44.4	50.0	55.6	61.1	66.7	72.2	77.8	83.3	88.9	94.4	100.0			
2.00	5.0	10.0	15.0	20.0	25.0	30.0	35.0	40.0	45.0	50.0	55.0	60.0	65.0	70.0	75.0	80.0	85.0	90.0	95.0	100.0	
3.00	3.3	6.7	10.0	13.3	16.7	20.0	23.3	26.7	30.0	33.3	36.7	40.0	43.3	46.7	50.0	53.3	56.7	60.0	63.3	66.7	
4.00	2.5	5.0	7.5	10.0	12.5	15.0	17.5	20.0	22.5	25.0	27.5	30.0	32.5	35.0	37.5	40.0	42.5	45.0	47.5	50.0	
5.00	2.0	4.0	6.0	8.0	10.0	12.0	14.0	16.0	18.0	20.0	22.0	24.0	26.0	28.0	30.0	32.0	34.0	36.0	38.0	40.0	
6.00	1.7	3.3	5.0	6.7	8.3	10.0	11.7	13.3	15.0	16.7	18.3	20.0	21.7	23.3	25.0	26.7	28.3	30.0	31.7	33.3	
7.00	1.4	2.9	4.3	5.7	7.1	8.6	10.0	11.4	12.9	14.3	15.7	17.1	18.6	20.0	21.4	22.9	24.3	25.7	27.1	28.6	
8.00	1.3	2.5	3.8	5.0	6.3	7.5	8.8	10.0	11.3	12.5	13.8	15.0	16.3	17.5	18.8	20.0	21.3	22.5	23.8	25.0	
9.00	1.1	2.2	3.3	4.4	5.6	6.7	7.8	8.9	10.0	11.1	12.2	13.3	14.4	15.6	16.7	17.8	18.9	20.0	21.1	22.2	
10.00	1.0	2.0	3.0	4.0	5.0	6.0	7.0	8.0	9.0	10.0	11.0	12.0	13.0	14.0	15.0	16.0	17.0	18.0	19.0	20.0	
12.00	0.8	1.7	2.5	3.3	4.2	5.0	5.8	6.7	7.5	8.3	9.2	10.0	10.8	11.7	12.5	13.3	14.2	15.0	15.8	16.7	
14.00	0.7	1.4	2.1	2.9	3.6	4.3	5.0	5.7	6.4	7.1	7.9	8.6	9.3	10.0	10.7	11.4	12.1	12.9	13.6	14.3	
16.00	0.6	1.3	1.9	2.5	3.1	3.8	4.4	5.0	5.6	6.3	6.9	7.5	8.1	8.8	9.4	10.0	10.6	11.3	11.9	12.5	
18.00	0.6	1.1	1.7	2.2	2.8	3.3	3.9	4.4	5.0	5.6	6.1	6.7	7.2	7.8	8.3	8.9	9.4	10.0	10.6	11.1	
20.00	0.5	1.0	1.5	2.0	2.5	3.0	3.5	4.0	4.5	5.0	5.5	6.0	6.5	7.0	7.5	8.0	8.5	9.0	9.5	10.0	
22.00	0.5	0.9	1.4	1.8	2.3	2.7	3.2	3.6	4.1	4.5	5.0	5.5	5.9	6.4	6.8	7.3	7.7	8.2	8.6	9.1	
24.00	0.4	0.8	1.3	1.7	2.1	2.5	2.9	3.3	3.8	4.2	4.6	5.0	5.4	5.8	6.3	6.7	7.1	7.5	7.9	8.3	
26.00	0.4	0.8	1.2	1.5	1.9	2.3	2.7	3.1	3.5	3.8	4.2	4.6	5.0	5.4	5.8	6.2	6.5	6.9	7.3	7.7	
28.00	0.4	0.7	1.1	1.4	1.8	2.1	2.5	2.9	3.2	3.6	3.9	4.3	4.6	5.0	5.4	5.7	6.1	6.4	6.8	7.1	
30.00	0.3	0.7	1.0	1.3	1.7	2.0	2.3	2.7	3.0	3.3	3.7	4.0	4.3	4.7	5.0	5.3	5.7	6.0	6.3	6.7	
32.00	0.3	0.6	0.9	1.3	1.6	1.9	2.2	2.5	2.8	3.1	3.4	3.8	4.1	4.4	4.7	5.0	5.3	5.6	5.9	6.3	
34.00	0.3	0.6	0.9	1.2	1.5	1.8	2.1	2.4	2.6	2.9	3.2	3.5	3.8	4.1	4.4	4.7	5.0	5.3	5.6	5.9	
36.00	0.3	0.6	0.8	1.1	1.4	1.7	1.9	2.2	2.5	2.8	3.1	3.3	3.6	3.9	4.2	4.4	4.7	5.0	5.3	5.6	
40.00	0.3	0.5	0.8	1.0	1.3	1.5	1.8	2.0	2.3	2.5	2.8	3.0	3.3	3.5	3.8	4.0	4.3	4.5	4.8	5.0	
44.00	0.2	0.5	0.7	0.9	1.1	1.4	1.6	1.8	2.0	2.3	2.5	2.7	3.0	3.2	3.4	3.6	3.9	4.1	4.3	4.5	
48.00	0.2	0.4	0.6	0.8	1.0	1.3	1.5	1.7	1.9	2.1	2.3	2.5	2.7	2.9	3.1	3.3	3.5	3.8	4.0	4.2	
52.00	0.2	0.4	0.6	0.8	1.0	1.2	1.3	1.5	1.7	1.9	2.1	2.3	2.5	2.7	2.9	3.1	3.3	3.5	3.7	3.8	
56.00	0.2	0.4	0.5	0.7	0.9	1.1	1.3	1.4	1.6	1.8	2.0	2.1	2.3	2.5	2.7	2.9	3.0	3.2	3.4	3.6	
60.00	0.2	0.3	0.5	0.7	0.8	1.0	1.2	1.3	1.5	1.7	1.8	2.0	2.2	2.3	2.5	2.7	2.8	3.0	3.2	3.3	

Example: When measuring 0.30 bar ppO<sub>2</sub> at a pressure of 1.40 bara, the oxygen comprises 21.4% of the atmosphere being measured.

### 3.7 Concerned whether reading is accurate?

If you are in any doubt as to the accuracy of the analyser reading, refer to Section 6.2 for guidance on how to check the credibility of the reading. If the reading is believed to be incorrect, then the analyser should be recalibrated as detailed in Section 4.

### 3.8 Effect of temperature changes

The oxygen sensor within the analyser is fitted with temperature compensation components which automatically adjust the output of the sensor at different temperatures. The user therefore ought not to be concerned as long as the temperature is within the normal operating temperature range. However, the compensation is not perfect, and thus it is necessary to allow for an error in measurement when working at different temperatures. We can't specify this precisely, but we can state the typical effect on the reading with changes in temperature (see Section 8). Please note that this figure represents temperature changes which have been allowed to stabilise. Performance will be slightly poorer during temperature transitions. The user should note whether the analyser is subjected to temperature changes, and therefore assess whether there is a need to allow a settling time before taking a reading. This can be particularly important when the analyser is cold and it is taken into a warm environment to take a measurement, or the reverse situation, when the analyser is already warm and it is taken into a cold environment.

### 3.9 Rapid changes of pressure

The analyser will withstand a wide range of operating pressure, but it must not be exposed to rapid changes of pressure. It will withstand rates of change which are accepted for humans (see Section 8).

Do not pass the analyser through chamber medical locks in which the pressure is rapidly changed.

### 3.10 Effect of humidity

As stated earlier, the analyser measures the partial pressure of oxygen. The sensor will continue to operate in high levels of humidity. The gas inlet to the oxygen sensor is protected by a hydrophobic membrane, which prevents passage of liquid water into the sensor. Care must be exercised that condensed water is not allowed to settle in the gas inlet port. A layer of water will block gas diffusion into the sensor. In practice this problem is easily overcome. Simply holding or mounting the analyser in a vertical rather than horizontal position will mean that any water collecting in the sensor inlet will simply run off the sensor.

So does humidity affect the reading? As an analyser measuring partial pressure of oxygen, the answer is effectively 'no'. However in many cases, the analyser may be used to measure atmospheres that are regarded as being 'air'. Care has to be exercised here, since in air, as the humidity increases, it displaces nitrogen and oxygen (and any other more minor constituents). Therefore, if at the time of calibration (calibration in air only), you have not accounted for humidity, then there may be a problem with the later oxygen readings. The effect is greatest for high levels of humidity at high temperatures. At low temperature or humidity, the effect is virtually insignificant.

As an example, if the analyser were calibrated at 38°C and a relative humidity of 90% as per Section 4.4, then the actual percentage of oxygen in a fresh air environment will be 19.7%. If the pressure is 1bar, then the partial pressure to which the analyser should be adjusted will be 0.20bar. Ignoring humidity, the user may have assumed that the analyser should be calibrated to 20.9% of 1bar (0.20bar). This would result in a gain error of approximately 6% of reading on all subsequent oxygen readings.

Conversely, if the same calibration were performed at 20% relative humidity and 10°C, then the oxygen content should be assumed as 20.8% rather than 20.9%. However, failure to account for the humidity in this example would result in a gain error of approximately 0.5% of reading which to all intents and purposes will usually be insignificant.

## 4 Calibration

### 4.1 Frequency of check/calibration

The analyser must be calibrated in order to provide accurate oxygen measurement. The analyser is calibrated at the factory immediately prior to shipping.

The oxygen sensor used inside the analyser degrades as it is consumed over its approximate 3 year life. This characteristic makes it important to periodically calibrate the sensor to compensate for the slow degradation taking place and hence maintain the analyser in a condition where it will provide adequate measurement accuracy.

The typical degradation is a decline of approximately 5% of reading per year.

So an analyser reading 0.21bar immediately after calibration might be expected to read 0.20bar one year later in exactly the same conditions (oxygen, temperature and pressure).

This has to be quantified against the user's requirements for how accurately the measurements are to be made.

Calibrating every 6 months would reduce the typical error above from 0.01bar down to 0.005bar which for most users is expected to be good enough. In some critical applications, some users insist that the analyser is calibrated every day. In this instance the daily 'calibration' is essentially little more than a 'check' that the instrument is still working normally.

## 4.2 Type of calibration

The instrument will most likely be calibrated in one of two ways depending on which best suits the intended use.

For users who need to make oxygen measurements over the entire range of the sensor, it is best to calibrate using pure oxygen.

For users who only expect to use the lower parts of the range (for instance in a submarine kept essentially near to normal atmospheric pressure and in air equivalent atmospheres), it may be preferable (for simplicity rather than accuracy) to simply perform an air calibration. In its simplest form this requires no specialist calibration gas, and is hence favoured by some users. However it is strictly not advisable if the analyser is then expected to make accurate measurements at significantly higher concentrations of oxygen. It also introduces further uncertainty into the calibration particularly at higher temperature/humidity. Whilst everyone expects that fresh air contains 20.9% oxygen, as relative humidity levels increase, and particularly as the temperature also increases allowing the atmosphere to carry more water vapour, then the actual amount of oxygen in the atmosphere is diluted. This can be accounted for, but to be done properly does require knowledge of both temperature and relative humidity. It is also possible to use a calibration gas, say calibration air. This overcomes the humidity problem, since the calibration gas will be dry. The problem with 20.9% calibration gas, is that its accuracy at best is 1% of reading, and generally 2% or even 5% for some suppliers.

So with a 1% accurate cal gas, a 20.9% oxygen may be anything from 20.7% up to 21.1%  
and a 2% accurate cal gas anything from 20.5% up to 21.3%  
and a 5% accurate cal gas anything from 18.9% up to 21.9%

If you calibrate with say the 5% accurate gas, and leave a residual error of 1% oxygen at an air concentration, you will find that if you then subject the sensor to pure oxygen (5 times the concentration), the error will have increased to 5% oxygen. So you may believe by measurement that you are breathing 95% oxygen when in fact it is perfectly pure.

By calibrating with pure oxygen, we overcome all of these problems. Firstly, pure oxygen calibration gas can be obtained to a degree of accuracy well beyond what the sensor can measure. You could use zero grade oxygen which is generally specified to be at least 99.5% pure, or you could go further and purchase oxygen with a purity of 99.995%, which in honesty is not generally worth the extra cost for this particular analyser.

As with the air cal gas, the pure oxygen will be dry, so we don't have to worry about humidity. But also, even using the 99.5% pure gas, we may have a 0.5% error in oxygen measurement, but as we now expose the sensor to air at approximately a fifth of the calibration gas concentration, the error in the air reading due to that initial 0.5% error will now be only 0.1%. This is effectively negligible in virtually all applications.

So which is best?

Without a doubt, calibration in pure oxygen is better as it can minimise all of the errors. But as expressed above, you can still obtain meaningful measurements using a fresh air calibration, so long as you account for temperature and humidity. Users would generally only resort to an air calibration gas in situations where it was not possible to use pure oxygen (for safety reasons) and because they do not need to measure such high levels of oxygen.

#### 4.3 Air calibration using fresh-air without temperature or humidity compensation

In this section, it is assumed that the fresh air in the atmosphere contains 20.9% oxygen. Do not use this mode of calibration in any confined space where the level of oxygen may be slightly depleted. This is the least accurate, but simplest means of calibration.

1	Switch the analyser on
2	Subject the sensor to fresh air. If the analyser was previously located at a temperature different to the fresh air, then leave the analyser for a few minutes to settle to a steady reading
3	Calculate the expected reading. Expected reading = $0.209 \times \text{Atmospheric Pressure (bar)}$
4	Use a small screwdriver to adjust the potentiometer accessible through the small hole on the front face of the analyser. Clockwise rotation will increase the reading. Adjust the reading to be as calculated above

#### 4.4 Air calibration using fresh-air with temperature and humidity compensation

In this section, it is assumed that the fresh air in the atmosphere contains slightly less than 20.9% oxygen, due to dilution with water vapour. Do not use this mode of calibration in any confined space where the level of oxygen may be slightly depleted.

1	Switch the analyser on
2	Subject the sensor to fresh air. If the analyser was previously located at a temperature different to the fresh air, then leave the analyser for a few minutes to settle to a steady reading
3	Obtain the temperature of the environment in either Fahrenheit or Centigrade to the nearest 5°C or 10°F. Refer to Table 3 and select the column in the table corresponding to the temperature.
4	Obtain the relative humidity of the environment in %RH, and select the row in Table 3 corresponding to this humidity.
5	Locate the cell in Table 3 where the selected row and column intersect. This will tell you the expected percentage of oxygen in the atmosphere.
6	Calculate the expected ppO <sub>2</sub> bar reading. Expected reading = $\text{Value from Table 3} \times \text{Atmospheric Pressure (bar)} / 100$ or alternatively use Table 4 to lookup the ppO <sub>2</sub> in bar from the %O <sub>2</sub> and the atmospheric pressure.
7	Use a small screwdriver to adjust the potentiometer accessible through the small hole on the front face of the analyser. Clockwise rotation will increase the reading. Adjust the reading to be as calculated above

**Table 3 : Humidity Compensation Chart**

Atmosphere oxygen percent in relation to temperature and relative humidity										
Temp °F	32	40	50	60	70	80	90	100	110	120
Temp °C	0	4	10	16	21	27	32	38	43	49
%RH	Atmospheric Oxygen Percent									
0	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9	20.9
10	20.9	20.9	20.9	20.9	20.8	20.8	20.8	20.8	20.7	20.7
20	20.9	20.9	20.8	20.8	20.8	20.8	20.7	20.6	20.5	20.4
30	20.9	20.8	20.8	20.8	20.7	20.7	20.6	20.5	20.4	20.2
40	20.8	20.8	20.8	20.7	20.7	20.6	20.5	20.4	20.2	19.9
50	20.8	20.8	20.8	20.7	20.6	20.5	20.4	20.2	20.0	19.7
60	20.8	20.8	20.7	20.7	20.6	20.5	20.3	20.1	19.8	19.5
70	20.8	20.8	20.7	20.6	20.5	20.4	20.2	19.9	19.6	19.2
80	20.8	20.8	20.7	20.6	20.5	20.3	20.1	19.8	19.5	19.0
90	20.8	20.7	20.7	20.6	20.4	20.3	20.0	19.7	19.3	18.7
100	20.8	20.7	20.6	20.5	20.4	20.2	19.9	19.5	19.1	18.5
%H2O at 100% RH	0.6	0.8	1.2	1.8	2.5	3.4	4.7	6.5	8.6	11.5
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**Table 4: ppO<sub>2</sub> in bar for given %O<sub>2</sub> at various atmospheric pressures**

	Atmospheric pressure (bar)						
	0.94	0.96	0.98	1.00	1.02	1.04	1.06
%O <sub>2</sub>							
19.0	0.18	0.18	0.19	0.19	0.19	0.20	0.20
19.1	0.18	0.18	0.19	0.19	0.19	0.20	0.20
19.2	0.18	0.18	0.19	0.19	0.20	0.20	0.20
19.3	0.18	0.19	0.19	0.19	0.20	0.20	0.20
19.4	0.18	0.19	0.19	0.19	0.20	0.20	0.21
19.5	0.18	0.19	0.19	0.20	0.20	0.20	0.21
19.6	0.18	0.19	0.19	0.20	0.20	0.20	0.21
19.7	0.19	0.19	0.19	0.20	0.20	0.20	0.21
19.8	0.19	0.19	0.19	0.20	0.20	0.21	0.21
19.9	0.19	0.19	0.20	0.20	0.20	0.21	0.21
20.0	0.19	0.19	0.20	0.20	0.20	0.21	0.21
20.1	0.19	0.19	0.20	0.20	0.21	0.21	0.21
20.2	0.19	0.19	0.20	0.20	0.21	0.21	0.21
20.3	0.19	0.19	0.20	0.20	0.21	0.21	0.22
20.4	0.19	0.20	0.20	0.20	0.21	0.21	0.22
20.5	0.19	0.20	0.20	0.21	0.21	0.21	0.22
20.6	0.19	0.20	0.20	0.21	0.21	0.21	0.22
20.7	0.19	0.20	0.20	0.21	0.21	0.22	0.22
20.8	0.20	0.20	0.20	0.21	0.21	0.22	0.22
20.9	0.20	0.20	0.20	0.21	0.21	0.22	0.22

4.5 Calibration using calibration gas

**SAFETY WARNING**

It is assumed that personnel using this procedure have been trained in the safe use of gas and associated equipment (gas bottles, regulators etc). Do not use high pressure gas if you have not been suitably trained.

To subject the analyser to a known concentration of calibration gas, use an Analox flow adaptor pushed into the sensor's gas inlet. Connect the flow adaptor either to a constant flow regulator (as shown in Figure 1) or to a flow meter in turn connected to a pressure regulator.

In either case, the aim is to achieve a flow of between 0.2 and 1.0 litres per minute of calibration gas through the flow adaptor.

The calibration gas will be certified as containing x% oxygen to an accuracy of y%. The accuracy (y%) is not required in the calibration, but it is important to recognise that this leads to an uncertainty in the resulting analyser reading. Typically y will be either 1, 2 or 5% depending on the grade of gas purchased.

The procedure could apply for all values of x in the range 1 to 100, but it would be very unusual to calibrate with anything less than 20% oxygen. The larger the number, the greater the accuracy of the analyser over its entire measuring range.

1	Switch the analyser on. If the analyser was previously located at a temperature different to the fresh air, then leave the analyser for a few minutes to settle to a steady reading
2	Subject the sensor to calibration gas at a flow rate of between 0.2 and 1.0 litres per minute. Allow the reading to settle which will take approximately one minute.
3	Calculate the expected reading. Expected reading = $(x / 100) \times \text{Atmospheric Pressure (bar)}$ where x is the certified percentage content of oxygen in the calibration gas
4	Use a small screwdriver to adjust the potentiometer accessible through the small hole on the front face of the analyser. Clockwise rotation will increase the reading. Adjust the reading to be as calculated above

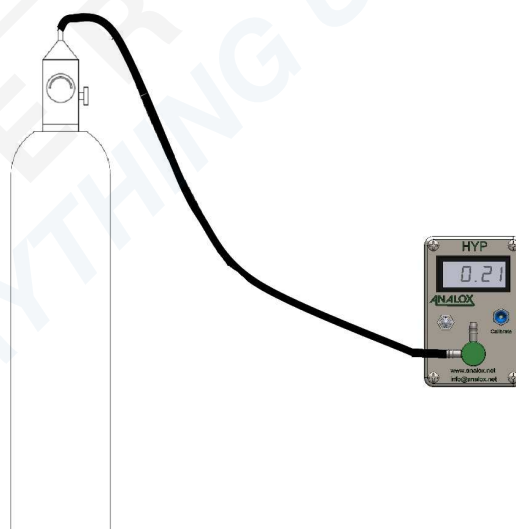


Figure 1 : Applying Calibration Gas using constant flow regulator

4.6 Calibration using pure oxygen

**SAFETY WARNING**

It is assumed that personnel using this procedure have been trained in the safe use of oxygen.  
Do not use oxygen if you have not been suitably trained.

This is essentially the same as in Section 4.5, but the value x will be between 99.5 and virtually 100% depending on the grade of oxygen purchased. Use exactly the same flow rate, but note there is an additional test after calibration that is actually a very good indication of the health of the sensor. Although this is the best method of calibration (if using the entire measuring range of the analyser), some users prefer not to use it because of the need to handle pure oxygen and the associated safety precautions necessary.

1	Switch the analyser on. If the analyser was previously located at a temperature different to the fresh air, then leave the analyser for a few minutes to settle to a steady reading
2	Subject the sensor to calibration gas (pure oxygen) at a flow rate of between 0.2 and 1.0 litres per minute. Allow the reading to settle which will take approximately one minute.
3	Calculate the expected reading. Expected reading = $(x / 100) \times \text{Atmospheric Pressure (bar)}$ where x is the certified percentage content of oxygen in the calibration gas – typically 99.5 or 99.9%
4	Use a small screwdriver to adjust the potentiometer accessible through the small hole on the front face of the analyser. Clockwise rotation will increase the reading. Adjust the reading to be as calculated above
5	Remove the flow adaptor from the sensor and allow the sensor to settle to the ambient air conditions.
6	Calculate the expected partial pressure reading for the ambient conditions. If this is fresh air at low levels of humidity, then this will be 20.9% oxygen. The expected reading will be Expected reading = $20.9 \times \text{Atmospheric Pressure (bar)} / 100$  For higher levels of humidity, obtain the expected percentage of oxygen by reference to Table 3, and substitute this for the 20.9 figure in the equation above.
7	Compare the analyser reading with the expected reading. It would be expected for the reading to be within +/-0.01bar of the expected reading, provided that the expected reading is not in error due to humidity effects or mis-calculation with the assumed atmospheric pressure.  At the end of the sensor's life, it will be found that this reading will be much higher than expected – perhaps as much as 0.02 to 0.04bar. The reason for this is that as the sensor ages, it will still measure low levels of oxygen (eg air atmospheres), but it will saturate and be unable to respond to high levels of oxygen (eg pure O <sub>2</sub> ). So what has happened is that the lower than expected reading for 100% O <sub>2</sub> has been artificially increased by use of the calibration potentiometer. This then correspondingly increases the apparent oxygen content when exposed to air.  The fact that the sensor cannot respond to higher levels of oxygen means the sensor is beyond the end of its useful life. Replace it at the earliest opportunity.

#### 4.7 Calibration against a reference analyser

In some circumstances, it is not possible to use fresh air or calibration gas, and yet there is still a need to check or calibrate the instrument. This can only be done if there is a reference available in which there is a suitably low uncertainty, such that it is deemed possible to calibrate the analyser against the reference.

It must be ensured that both the HYP O2 Portable and the reference are both measuring the same gas. Do not assume that simply because both analysers are in the same room that they ought to measure the same. Locate both analysers together, or subject them to the same sample gas.

If the reference is a partial pressure analyser itself, then simply adjust the HYP O2 Portable analyser to give the same reading (it may be necessary to convert from one set of units to another unless both analysers are reading in Torr).

If the reference analyser measures percentage of oxygen, convert this to a partial pressure by obtaining the local ambient pressure, preferably in bar. Then adjust the HYP O2 Portable analyser to give a reading equal to  $(\%O_2/100 \times \text{Ambient Pressure in Bar})$ .

## 5 Maintenance

### 5.1 Replacing battery

The analyser uses a single 9V PP3 type battery. If using the analyser at high pressure (>1.5 bar absolute), you may consider using a battery that has been tested to operate under hyperbaric conditions. These are available from Analox (refer Section 9). In truth, very few PP3 batteries are known to have failed under pressure, but in critical applications this attention to detail can minimise the risk of failure.

Simply unscrew the four screws securing the lid, then slide the battery from under its retaining clip which is mounted to the enclosure base.

Unclip it from the battery connector. Fit the new replacement battery to the battery connector and locate the battery under its retaining clip, before refitting the enclosure lid back into place. Ensure that the battery wiring does not become trapped in the process. Secure in place with the previously removed screws.

Discard the old battery in line with your own battery disposal procedures.

## 5.2 Replacing oxygen sensor

Please read the safety information in Section 7 before handling oxygen sensors.

The analyser uses an Analox supplied oxygen sensor. The sensor will typically last for 3 years when exposed to normal air at normal atmospheric pressure. Regular use in either high temperature environments or in situations using partial pressures of oxygen greater than 209 mbar ppO<sub>2</sub> (0.21bar ppO<sub>2</sub>) will cause the sensor to deteriorate more rapidly, and hence shorten its life. For instance, exposure to pure oxygen at standard atmospheric pressure would effectively consume the oxygen sensor approximately five times faster than normal. If maintained in this atmosphere continuously, the sensor would need replacing typically every 6 months. However such exposure would represent fairly severe circumstances, and is not considered likely.

Most commercial diving and military customers tend to replace the sensor at 12, 18 or 24 months, depending on their mode of usage.

To replace the sensor:

1	Switch the analyser off
2	Remove the four cross-head screws from the enclosure lid
3	Separate the two halves of the enclosure.
4	Remove the sensor from the enclosure lid, this should just pull out
5	Disconnect the Molex two pin connector that connects the oxygen sensor to the printed circuit board
6	Fit the new oxygen sensor to the front face of the enclosure by pushing it through the larger hole from the underside
7	Connect the sensor wiring lead to the two pin connector
8	Ensure that the wiring is neat and tidy and will not become trapped when re-assembling the enclosure
9	Refasten the enclosure using the 4 screws

## 6 Troubleshooting

### 6.1 Instrument will not switch on

Check that the battery is correctly inserted and that the battery is not exhausted.

Check that the battery wiring has not become damaged due to being trapped in the case. Refer to the instructions in Section 5.1 about changing the battery and ensuring the wiring is not trapped.

### 6.2 Instrument provides an oxygen reading that is believed to be incorrect

If in doubt about the oxygen measurement, it is best, if possible, to expose the analyser to a known oxygen concentration, which will give a good indication whether the instrument is working correctly or not. In many cases, a normal fresh air atmosphere is a readily available reference, although it is still necessary to know the atmospheric pressure to work out the actual partial pressure of oxygen. If fresh air is not available (for example in a submarine), then the analyser reading should be compared with that from another measuring device (refer Section 4.7), or checked against a bottle of calibration gas (refer Section 4.5).

### 6.3 Instrument will not calibrate

If during calibration the calibration potentiometer does not provide sufficient adjustment of the readout, check first that the calibration conditions are correct. Are you trying to adjust to the correct figure for the gas to which the sensor is exposed? Ultimately, if the sensor is not replaced at the correct service intervals, an expired sensor will give too low a reading, and it will become impossible to adjust the reading to the desired value. Correct maintenance ought to ensure that this condition is never encountered, because if the analyser is required for immediate use, it will be necessary to fit a new sensor first.

## 7 Safety information

The oxygen sensor is an electrochemical device and contains a caustic electrolyte. Always check to make sure that it is not leaking and do not allow it onto any part of your body or clothing.

When the life of the sensor has expired or it is leaking or otherwise damaged it must be disposed of safely in accordance with local regulations.

The electrolyte is potassium hydroxide solution (KOH) which is hazardous.

In the event of an accident, use the following first aid procedures

Body Part	Effect	First Aid Procedures
Skin	Contact could result in a chemical burn. Persons with pre-existing skin disorders may be more susceptible to the effects of the substance.	Immediately flush the skin thoroughly with water for at least 15 minutes. Remove contaminated clothing and wash before re-use. Obtain medical advice if continued irritation.
Ingestion	Corrosive. May cause sore throat, abdominal pain, nausea, and severe burns of the mouth, throat, and stomach, and may be fatal.	If swallowed DO NOT INDUCE VOMITING. Wash out mouth thoroughly with water and give plenty of water to drink. Obtain medical advice immediately
Eye	Persons with pre-existing eye problems may be more susceptible to the effects of the substance.  Corrosive. May cause redness, pain, blurred vision, and eye burns.  Contact can result in the permanent loss of sight.	Irrigate thoroughly with water for at least 15 minutes. Obtain medical advice immediately.
Inhalation	Persons with pre-existing impaired respiratory function may be more susceptible to the effects of the substance. Inhalation is not an expected hazard unless heated to high temperatures. Mist or vapour inhalation can cause irritation to the nose, throat, and upper respiratory tract.	Remove to fresh air. Rest and keep warm. Obtain medical advice if applicable.

## 8 Specifications

Sensor type	Analox 9100- 9212-5H
Sensor life	Up to 3 years in air at standard atmospheric pressure. Recommend replacement at 12,18 or 24 months dependant upon type of use.
Range	0 to 2bar ppO <sub>2</sub>
Accuracy	± 1% of reading plus ± 1LSD
Temperature effect	± 0.2% of reading per °C
Long term drift	typically 5% of reading per year
Battery type	9V PP3
Battery life	1 month (800 hours) continuous operation 3 months operation for 8 hours per day
Operating temperature range	0 to 40 °C
Operating pressure range	0.8 to 60bar absolute. (approx 600 MSW / 2000FSW)
Rate of compression/decompression	< 2 bar /minute
Dimensions	64mm (width) x 98mm (height) x 36mm (depth)
Weight	0.200kg
Warranty	Electronics – 1 year Sensor – 3 months

**9 Spares**

9100-9212-5H	Replacement oxygen sensor
2557-1855	Hyperbarically tested PP3 battery
GENPISDK	Sample Draw Kit, consisting of push in flow adaptor, aspirator bulb, 2m of neoprene tubing and a stainless steel probe.
SA7 100HYP CALKIT	Calibration kit comprising 110 litre 21% oxygen balance nitrogen 0.5 litre/minute flow regulator to fit 110 litre bottle push in flow adaptor 1 metre of tubing trimming tool
8000-0011GA	Flow adaptor (note not required if you already have either of the above kits)

## 10 Disposal



According to WEEE regulation this electronic product can not be placed in household waste bins. Please check local regulations for information on the disposal of electronic products in your area.

11 Declaration of conformity

**DECLARATION OF CONFORMITY**

Number: MO2-913-00

Manufacturers name: Analox Sensor Technology Ltd

Manufacturers address: 15 Ellerbeck Court  
Stokesley Business Park  
North Yorkshire, UK  
TS9 5PT

It is declared that the following product:


Product name: Analox Mini O2 HYP

Conforms to all applicable requirements of: BS EN 50270:1999,  
BS EN 61000-6-3:2001,

The above product complies with the requirements of the EMC Directive 89/336/EEC, as amended.

Signed on behalf of: Analox Sensor Technology Ltd

Date: 20<sup>th</sup> Dec 2012

Signed: 

Name: Mark Lewis

Position: Managing Director

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Appendix A Conversion Tables

Table 5: Convert Dew point in °C to Relative humidity in %

Dew Point (°C)	Ambient Temperature (°C)									
	0	4	10	16	21	27	32	38	43	49
-50	1	0	0	0	0	0	0	0	0	0
-25	10	8	5	3	3	2	1	1	1	1
-22	14	10	7	5	3	2	2	1	1	1
-18	20	15	10	7	5	4	3	2	1	1
-14	30	22	15	10	7	5	4	3	2	2
-11	39	29	19	13	10	7	5	4	3	2
-8	51	38	25	17	12	9	7	5	4	3
-6	60	45	30	20	15	10	8	6	4	3
-4	72	54	36	24	18	12	9	7	5	4
-2	85	64	42	28	21	15	11	8	6	4
-1	92	69	46	31	23	16	12	8	7	5
0	100	75	50	34	25	17	13	9	7	5
1		81	54	36	26	18	14	10	8	6
3		93	62	42	30	21	16	11	9	6
4		100	66	45	33	23	17	12	9	7
5			71	48	35	24	18	13	10	7
7			82	55	40	28	21	15	12	9
9			93	63	46	32	24	17	13	10
10			100	68	49	34	26	19	14	10
12				77	56	39	29	21	16	12
14				88	64	45	34	24	18	14
16				100	73	51	38	27	21	15
18					83	58	43	31	24	18
20					94	66	49	35	27	20
21					100	70	52	38	29	21
23						79	59	42	32	24
25						89	67	48	37	27
27						100	75	54	41	30
28							79	57	44	32
30							89	64	49	36
32							100	72	55	40
34								80	62	45
36								90	69	51
38								100	77	56
39									81	60
41									90	66
43									100	74
45										82
47										90
49										100

